

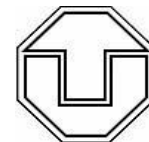


Thin Film Deposition

Prof. Ana L. F. de Barros

Centro Federal de Educação Tecnológica /CEFET-RJ,

Rio de Janeiro, RJ, Brazil.



TECHNISCHE
UNIVERSITÄT
DRESDEN



Thin Film Deposition

- ▶ General Characteristics
- ▶ Evaporation
- ▶ Langmuire–Knudsen Relations
- ▶ Uniform Coating
- ▶ Uniformity on a Flat Surface
- ▶ Thickness Deposition

II. Thin Film Deposition

Physical Vapor Deposition (PVD)

- Film is formed by atoms directly transported from source to the substrate through gas phase

- Evaporation
 - Thermal evaporation ↪
 - E-beam evaporation ↪
- Sputtering
 - DC sputtering ↪
 - DC Magnetron sputtering ↪
 - RF sputtering ↪
- Reactive PVD

Chemical Vapor Deposition (CVD)

- Film is formed by chemical reaction on the surface of substrate

- Low-Pressure CVD (LPCVD) ↪
- Plasma-Enhanced CVD (PECVD) ↪
- Atmosphere-Pressure CVD (APCVD)
- Metal-Organic CVD (MOCVD)

Oxidation

Spin Coating

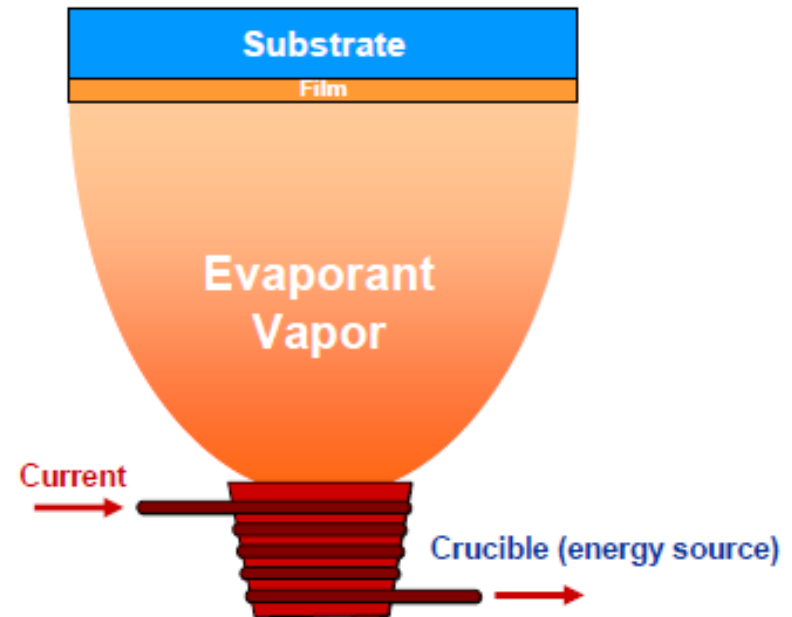
Plating

General Characteristics of Thin Film Deposition

- **Deposition Rate**
- **Film Uniformity**
 - Across wafer uniformity
 - Run-to-run uniformity
- **Materials that can be deposited**
 - Metal
 - Dielectric
 - Polymer
- **Quality of Film – Physical and Chemical Properties**
 - Stress
 - Adhesion
 - Stoichiometry
 - Film density, pinhole density
 - Grain size, boundary property, and orientation
 - Breakdown voltage
 - Impurity level
- **Deposition Directionality**
 - Directional: good for lift-off, trench filling
 - Non-directional: good for step coverage
- **Cost of ownership and operation**

Evaporation

- ➔ Load the source material-to-be-deposited (evaporant) into the container (crucible)
- ➔ Heat the source to high temperature
- ➔ Source material evaporates
- ➔ Evaporant vapor transports to and impinges on the surface of the substrate
- ➔ Evaporant condenses on and is adsorbed by the surface



Langmuire-Knudsen Relation

Mass Deposition Rate per unit area of source surface:

$$R_m = C_m \left(\frac{M}{T} \right)^{\frac{1}{2}} \cos \theta \cos \varphi \frac{1}{r^2} (P_e(T) - P)$$

$$C_m = 1.85 \times 10^{-2}$$

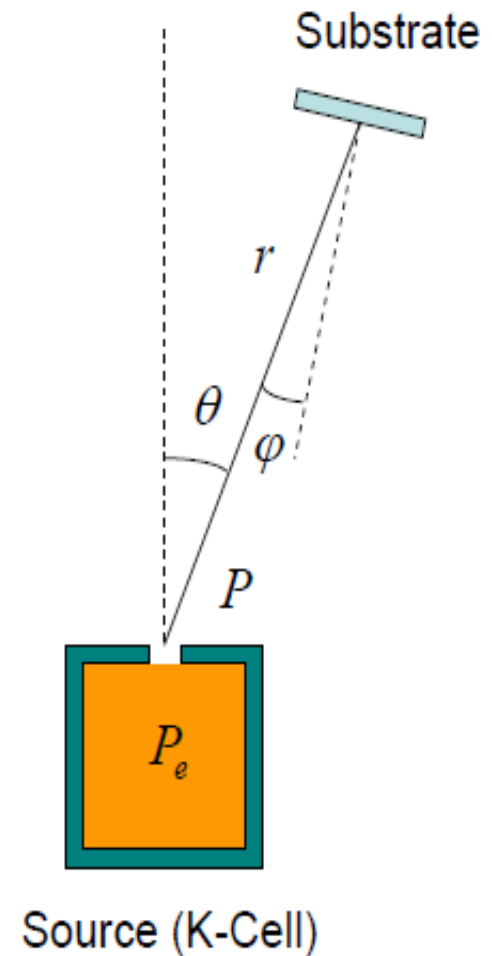
r : source-substrate distance (cm)

T : source temperature (K)

P_e : evaporant vapor pressure (torr), function of T

P : chamber pressure (torr)

M : evaporant gram-molecular mass (g)



➔ **Maximum deposition rate reaches at high chamber vacuum ($P \sim 0$)**

Uniform Coating

Spherical surface with source on its edge:

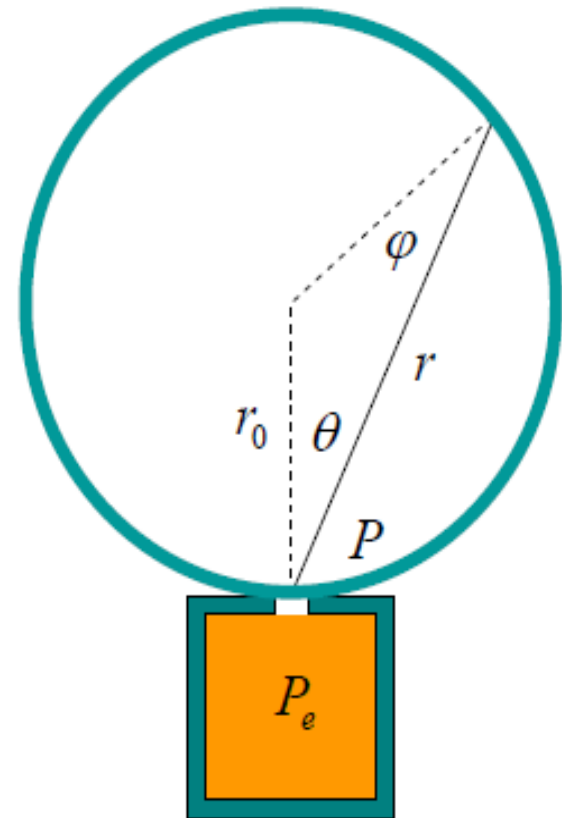
$$\cos \theta = \cos \varphi = \frac{r}{2r_0}$$

$$R_m = C_m \left(\frac{M}{T} \right)^{\frac{1}{2}} \frac{P_e}{4r_0^2}$$

➔ Angle Independent – uniform coating!

➔ Used to coat instruments with spherical surfaces

Spherical Surface



Source (K-Cell)

Uniformity on a Flat Surface

Consider the deposition rate difference between wafer center and edge:

$$R_1 \propto \frac{1}{r_1^2}$$

$$R_2 \propto \frac{1}{r_2^2} \cos^2 \theta = \frac{r_1^2}{r_2^4}$$

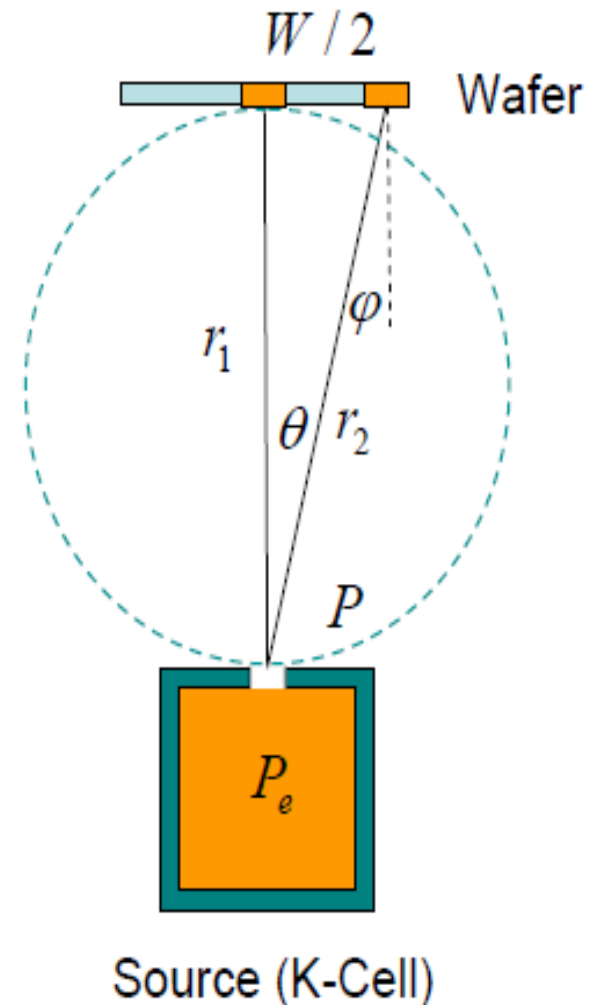
Define Uniformity:

$$\sigma(\%) = \frac{R_1 - R_2}{R_1} (\%)$$

$$\sigma = 1 - \left(1 + \left(\frac{W}{2r_1} \right)^2 \right)^{-2} \approx \frac{W^2}{2r_1^2}$$

or

$$\frac{W}{r_1} = \sqrt{2\sigma}$$



Uniformity Requirement on a Flat Surface

Source-substrate distance requirement:

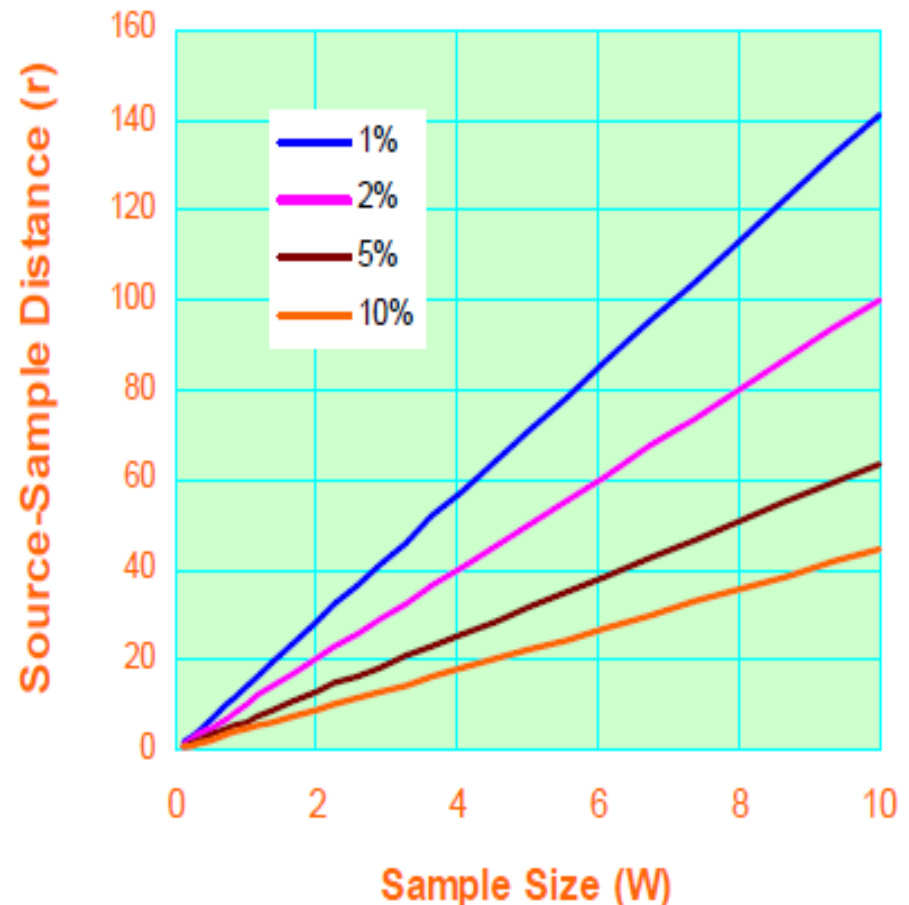
$$r > \frac{W}{\sqrt{2}\sigma}$$

In practice, it is typical to double this number to give some process margin:

$$r > W \sqrt{\frac{2}{\sigma}}$$

Larger r Means:

- ➔ bigger chamber
- ➔ higher capacity vacuum pump
- ➔ lower deposition rate
- ➔ higher evaporant waste



Another Common Solution:
off-axis rotation of the sample

Thickness Deposition Rate vs. Source Vapor Pressure

Thickness deposition rate

$$\frac{dh}{dt} = \frac{R_m}{\rho} A_e$$

$$\frac{dh}{dt} = \frac{A_e}{\rho} C_m \left(\frac{M}{T} \right)^{\frac{1}{2}} \cos \theta \cos \varphi \frac{1}{r^2} P_e(T)$$

T : source temperature (K)

A_e : source surface area (cm^2)

ρ : evaporant density (g/cm^3)

P_e is function of source Temperature!

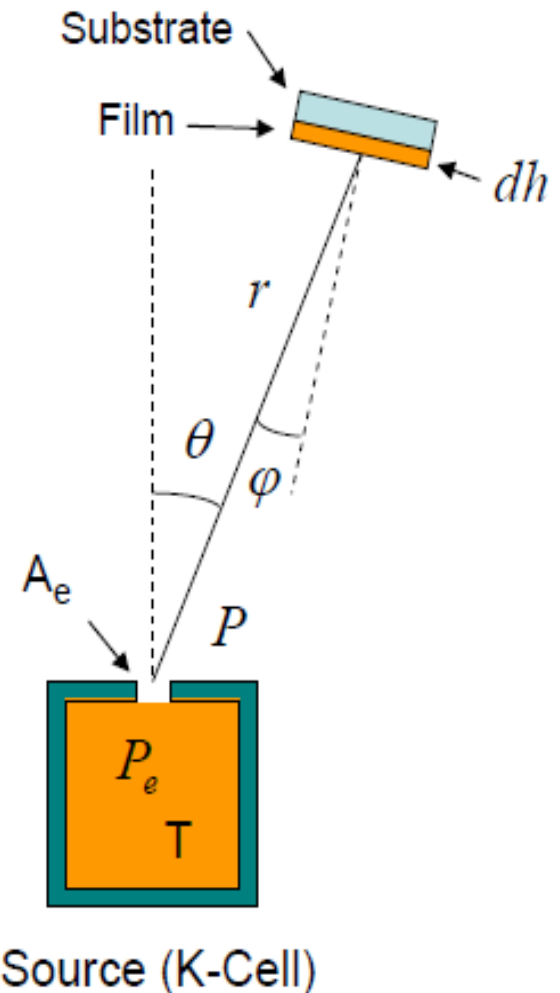
Example: Al

$M \sim 27$, $\rho \sim 2.7$, $A_e \sim 10^{-2} \text{ cm}^2$, $T \sim 900 \text{ K}$

$R \sim 50 \text{ cm}$ (uniformity requirement)

$$\frac{dh}{dt} = 50 P_e \quad (\text{\AA}/\text{s})$$

→ The higher the vapor pressure, the higher the material's deposition rate

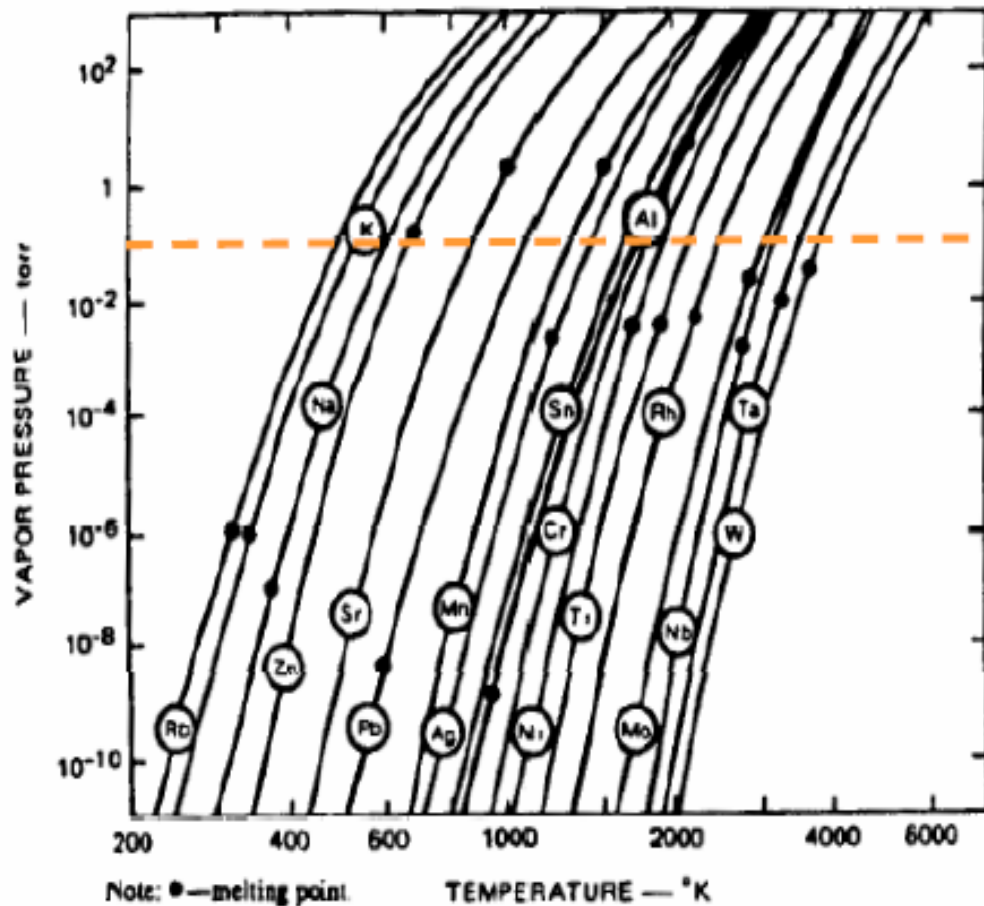


Deposition Rate vs. Source Temperature

Typically for different material:

$$\frac{dh}{dt} = (10 \sim 100)P_e(T) \quad (A/s)$$

- For deposition rate > 1 A/s:
 $P_e > \sim 100$ mtorr
- P_e depends on: 1) material and 2) temperature
- Deposition rates are significantly different for different materials
- Hard to deposit multi-component (alloy) film without losing stoichiometry



Example: for $P_e > 100$ mtorr

$T(\text{Al}) > 1400\text{K}$, $T(\text{Ta}) > 2500\text{K}$